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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/721,615	11/24/2003	Kenichiro Ueda	NGW-014	5617
959 75	590 09/25/2006		EXAMINER	
LAHIVE & COCKFIELD 28 STATE STREET			LEWIS, BEN	
BOSTON, MA 02109			ART UNIT	PAPER NUMBER
		•	1745	
		DATE MAILED: 09/25/2006		

Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No.	Applicant(s)		
	10/721,615	UEDA ET AL.		
Office Action Summary	Examiner	Art Unit		
	Ben Lewis	1745		
The MAILING DATE of this communication ap Period for Reply	pears on the cover sheet with the o	correspondence address		
A SHORTENED STATUTORY PERIOD FOR REPL WHICHEVER IS LONGER, FROM THE MAILING D - Extensions of time may be available under the provisions of 37 CFR 1. after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory period - Failure to reply within the set or extended period for reply will, by statute that the period for reply will, by statute that the mailing earned patent term adjustment. See 37 CFR 1.704(b).	OATE OF THIS COMMUNICATION 136(a). In no event, however, may a reply be tir will apply and will expire SIX (6) MONTHS from e, cause the application to become ABANDONE	N. nely filed the mailing date of this communication. D (35 U.S.C. § 133).		
Status				
Responsive to communication(s) filed on This action is FINAL2b) ☐ This Since this application is in condition for alloware closed in accordance with the practice under the practice under the practice.	s action is non-final. ance except for formal matters, pro			
Disposition of Claims				
4) ☐ Claim(s) 1-3 is/are pending in the application. 4a) Of the above claim(s) is/are withdra 5) ☐ Claim(s) is/are allowed. 6) ☐ Claim(s) 1-3 is/are rejected. 7) ☐ Claim(s) is/are objected to. 8) ☐ Claim(s) are subject to restriction and/o	wn from consideration.			
Application Papers				
9) The specification is objected to by the Examine 10) The drawing(s) filed on is/are: a) accomplicant may not request that any objection to the Replacement drawing sheet(s) including the correct 11) The oath or declaration is objected to by the Examine	cepted or b) objected to by the drawing(s) be held in abeyance. Section is required if the drawing(s) is ob	e 37 CFR 1.85(a). jected to. See 37 CFR 1.121(d).		
Priority under 35 U.S.C. § 119				
 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: 1. Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received. 				
Attachment(s) 1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date	4) Interview Summary Paper No(s)/Mail D 5) Notice of Informal F 6) Other:	ate		

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Detailed Action

1. The Applicant's amendment filed on July 5th, 2006 was received. Claim 1 was amended.

2. The text of those sections of Title 35, U.S.C. code not included in this action can be found in the prior Office Action (issued on April 4th, 2006).

Claim Rejections - 35 USC § 112

3. The following is a quotation of the first paragraph of 35 U.S.C. 112:

The specification shall contain a written description of the invention, and of the manner and process of making and using it, in such full, clear, concise, and exact terms as to enable any person skilled in the art to which it pertains, or with which it is most nearly connected, to make and use the same and shall set forth the best mode contemplated by the inventor of carrying out his invention.

Claim 1 is rejected under 35 U.S.C. 112, first paragraph, as failing to comply with the enablement requirement. The claim(s) contains subject matter which was not described in the specification in such a way as to enable one skilled in the art to which it pertains, or with which it is most nearly connected, to make and/or use the invention. It is not clear as to when hydrogen discharge can and cannot occur from the claimed limitation that "a hydrogen discharge from the fuel cell by the hydrogen discharge unit is prohibited in the event that the instantaneous hydrogen concentration does not exceed the first threshold but the average hydrogen concentration calculated by the average hydrogen concentration calculating unit exceeds a second threshold which is lower than the first threshold."

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Claim Rejections - 35 USC § 102

4. Claims 1-3 are rejected under 35 U.S.C. 102(e) as being anticipated by Yoshizumi et al. (U.S. Pub. No. 2002/0094469 A1).

With respect to claim 1, Yoshizumi et al disclose an onboard fuel cell system and method of discharging hydrogen-off gas wherein the fuel cell is supplied with hydrogen gas and oxidative gas, generates electric power using the hydrogen gas and the oxidative gas, and discharges hydrogen-off gas and oxygen-off gas that have been consumed (Paragraph 0011). The hydrogen gas that has thus been supplied is consumed for the electrochemical reactions in the fuel cell 100, turns into hydrogen-off gas, and is discharged to the circulation flow passage 403 (Paragraph 0052). Then, hydrogen gas that has been discharged from the shut valve 414 flows through the exhaust flow passage 407, is delivered to the oxygen-off gas exhaust flow passage 503, and is mixed with oxygen-off gas flowing through the oxygen-off gas exhaust flow passage 503 in the mixing portion 411. Because hydrogen gas discharged from the shut valve 414 is hydrogen-off gas, the concentration of hydrogen is relatively low. Also, oxygen-off gas discharged from the fuel cell 100 is also nitrogen-rich gas that has been removed of oxygen in the fuel cell 100. Accordingly, if hydrogen-off gas is thus mixed with oxygen-off gas and diluted, the concentration of hydrogen contained in the mixed gases is further reduced (Paragraph 0058). However, it is possible to dispose a

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hydrogen concentration sensor or the like in the hydrogen gas flow passage, detect the concentration of hydrogen in circulating hydrogen gas, and open the shut valve **414** if the concentration drops below a reference concentration (Paragraph 0057).

With regard to the hydrogen concentration calculating unit Yoshizumi et al teach that the shut valve **414** is disposed in the exhaust flow passage **407** that branches off from the circulation flow passage **403**. If the concentration of impurities in circulating hydrogen gas is increased, the control portion **50** opens the shut valve **414** so as to discharge part of the circulating hydrogen gas that contains impurities. Thereby, part of the hydrogen gas containing the impurities is discharged from the circulation passage, and pure hydrogen gas is introduced correspondingly from the hydrogen-occluding alloy tank 200 (Paragraph 0055).

With respect to claim 2, Yoshizumi et al teach that circulation of hydrogen-off gas can prevent impurities contained in oxidative gas such as oxygen and the like from piling up and that the output voltage of the fuel cell 100 can be prevented from dropping. The control portion 50 controls the driving of the pump 410, and the pump 410 changes the flow rate of hydrogen-off gas flowing through the circulation flow passage 403 in accordance with the amount of consumption of electric power generated by the fuel cell 100 (Paragraphs 0114 and 0115).

With respect to claim 3, Yoshizumi et al teach that shut valve **414** is disposed in the exhaust flow passage **407** that branches off from the circulation flow passage **403**.

If the concentration of impurities in circulating hydrogen gas is increased, the control portion 50 opens the shut valve 414 so as to discharge part of the circulating hydrogen gas that contains impurities. Thereby, part of the hydrogen gas containing the impurities is discharged from the circulation passage, and pure hydrogen gas is introduced correspondingly from the hydrogen-occluding alloy tank 200. Thus, the concentration of the impurities in hydrogen gas is reduced. Conversely, the concentration of hydrogen is increased. As a result, the fuel cell 100 can generate electric power continuously and suitably. Although the interval at which the shut valve 414 is opened differs depending on the operational condition or the output, the shut valve 414 may be opened, for example, at intervals of about 5 seconds.

Response to Arguments

5. Applicant's arguments filed on July 5th, 2006, have been fully considered but they are not persuasive.

Applicant's principle arguments are

(a) Yoshizumi reference fails to disclose the limitation that a hydrogen discharge from the fuel cell by the hydrogen discharge unit is prohibited in the event that the instantaneous hydrogen concentration does not exceed the first threshold; but the average hydrogen concentration

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calculating unit a second threshold which is lower than the first threshold, as recited in

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amended claim 1.

In response to Applicant's arguments, please consider the following comments.

(a) The claim(s) contains subject matter which was not described in the

specification in such a way as to enable one skilled in the art to which it pertains, or with

which it is most nearly connected, to make and/or use the invention. It is not clear as to

when hydrogen discharge can and cannot occur from the claimed limitation that " a

hydrogen discharge from the fuel cell by the hydrogen discharge unit is prohibited in the

event that the instantaneous hydrogen concentration does not exceed the first threshold

but the average hydrogen concentration calculated by the average hydrogen

concentration calculating unit exceeds a second threshold which is lower than the first

threshold."

Conclusion

6. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, THIS ACTION IS MADE FINAL. See MPEP

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§ 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Ben Lewis whose telephone number is 571-272-6481. The examiner can normally be reached on 8:30am - 5:30pm.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Patrick Ryan can be reached on 571-272-1292. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

Ben Lewis

PATRICK JOSEPH RYAN SUPERVISORY PATENT EXAMINER

Paten Examiner Art Unit 1745